

## ION NAMING WORKSHEET

Name: \_\_\_\_\_

Period: \_\_\_\_\_

### Part I – Name the following ions/elements

$\text{Na}^+$  \_\_\_\_\_

$\text{F}^-$  \_\_\_\_\_

$\text{Mg}$  \_\_\_\_\_

$\text{Be}^{2+}$  \_\_\_\_\_

$\text{O}^{2-}$  \_\_\_\_\_

$\text{P}^{3-}$  \_\_\_\_\_

$\text{Al}^{3+}$  \_\_\_\_\_

$\text{Cl}^-$  \_\_\_\_\_

$\text{Br}_2$  \_\_\_\_\_

$\text{Li}^+$  \_\_\_\_\_

### Part II – Write the formula – Be sure to include charges if the particle is an ion

Potassium ion \_\_\_\_\_

Bromide \_\_\_\_\_

Magnesium ion \_\_\_\_\_

Sulfide \_\_\_\_\_

Carbon \_\_\_\_\_

Cesium ion \_\_\_\_\_

Aluminum ion \_\_\_\_\_

Phosphide \_\_\_\_\_

### Part III – Give the formula or name of the following transition metal ions

Lead (IV) ion \_\_\_\_\_

$\text{Cu}^{2+}$  \_\_\_\_\_

$\text{Cu}^+$  \_\_\_\_\_

$\text{Ag}^+$  \_\_\_\_\_

Zinc ion \_\_\_\_\_

$\text{Fe}^{3+}$  \_\_\_\_\_

Manganese (II) ion \_\_\_\_\_

Titanium (III) ion \_\_\_\_\_

### Part IV – Give the formula or name of the following poly atomic ions

Phosphate \_\_\_\_\_

$\text{NO}_3^-$  \_\_\_\_\_

Sulfate \_\_\_\_\_

$\text{CO}_3^{2-}$  \_\_\_\_\_

Bicarbonate \_\_\_\_\_

$\text{NO}_2^-$  \_\_\_\_\_

Chlorate \_\_\_\_\_

$\text{NH}_4^+$  \_\_\_\_\_

Hydroxide \_\_\_\_\_

$\text{HSO}_3^-$  \_\_\_\_\_